## THE STRUCTURE OF CODONOPSININE

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Continuing an investigation of the epigeal part of Codonopsis clematidea (clematis asiabell) from the mother liquors from codonopsine (I), we have isolated a new base with the composition  $C_{13}H_{19}O_3N$ , mol. wt. 237 (mass spectrometry), mp 169-170°C (methanol),  $[\alpha]_D^{20}$ -8.8° (c 0.1; methanol),  $R_f$  0.23 [on a thin layer of silica gel-gypsum in ethyl acetate-methanol (9:1) system], and we have called it codonopsinine.

The IR spectrum of the alkaloid has absorption bands at (cm $^{-1}$ ): 812, 705 (disubstituted benzene nucleus) and 3375 (active hydrogen). UV spectrum,  $\lambda_{\max}^{C_2H_5OH}$ , m $\mu$ : 228, 277, and 284 (inflection) (log  $\epsilon$  4.14, 3.22, and 3.14, respectively). The IR and UV spectra of codonopsinine are similar to those for codonopsine.

The NMR spectrum of the base taken on a JNM-4H-100/100 MHz spectrometer in deuterochloroform show signals from the protons of a N-CH<sub>3</sub> group (singlet,  $\tau$ , 7.92), a CH-CH<sub>3</sub> group (doublet,  $\tau$  8.85), and an OCH<sub>3</sub> group (singlet,  $\tau$  6.46). A comparison of the NMR spectra of this base and of codonopsine showed that codonopsinine contains only one methoxyl group in the aromatic moiety.

The mass-spectrometric fragmentation of codonopsinine is similar to that of codonopsine [2]. The spectrum has peaks of ions with m/e 237 (M<sup>+</sup>), 177, 176, and 162, which differ from the corresponding peaks of codonopsine by 30 m/e.

The oxidation of codonopsinine with potassium permanganate in acetone yielded an acid with mp 110°C, which was identified by a mixed-melting-point test as m-methoxybenzoic acid.

On the basis of what has been said above, it is possible to propose for codonopsinine the structure 3, 4-dihydroxy-5-(m-methoxyphenyl)-1, 2-dimethyl-pyrrolidine:

## REFERENCES

- 1. S. F. Matkhalikova, V. M. Malikov, and S. Yu. Yunusov, KhPS [Chemistry of Natural Compounds], 5, 30, 1969.
- 2. S. F. Matkhalikova, V. M. Malikov, and S. Yu. Yunusov, KhPS [Chemistry of Natural Compounds], 5, 606, 1969 (in this issue.)

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